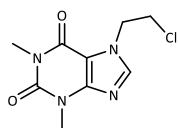


## SUPPLEMENTARY INFORMATION

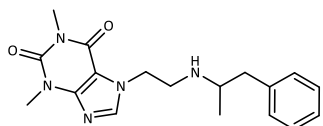
### Vaccine-driven pharmacodynamic dissection and mitigation of Captagon psychoactivity

**Chemical Analysis.** Reactions using microwave energy heating were run on an Explorer 12 Hybrid (CEM). Final chemical compounds synthesized in the course of this study were isolated at >95% purity following HPLC separation (5-95% ACN in H<sub>2</sub>O; 60 min; 10 mL/min) using an Agilent 1260 Infinity system and Vydac C18 column (4.6 x 250 mm). Nuclear magnetic resonance (<sup>1</sup>H NMR (500/400 MHz), <sup>13</sup>C NMR (125/100 MHz)) spectra were determined on either a Bruker 500 or 400 instrument. Chemical shifts for <sup>1</sup>H NMR are reported in parts per million (ppm) relative to chloroform-*d* (7.26 ppm), Methanol-*d*<sub>4</sub> (3.31 ppm), or DMSO-*d*<sub>6</sub> (2.50 ppm) and coupling constants are in hertz (Hz). The following abbreviations are used for spin multiplicity: s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, br = broad. Chemical shifts for <sup>13</sup>C NMR were reported in ppm relative to the signal at 77.16 ppm for chloroform-*d*, 49.00 for Methanol-*d*<sub>4</sub>, or 39.52 ppm for DMSO-*d*<sub>6</sub>. Electrospray Ionization (ESI) mass were obtained on a ThermoFinnigan LTQ Ion Trap. Matrix-assisted Laser Desorption/Ionization (MALDI) spectra were obtained on an Applied Biosystems Voyager DE.

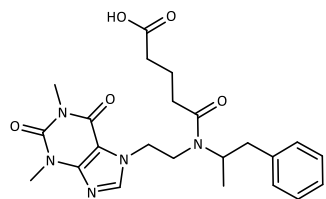
#### Synthesis of FEN Hapten.



**7-(2-chloroethyl)-1,3-dimethyl-3,7-dihydro-1H-purine-2,6-dione (7).** To a microwave vessel containing a solution of **6** (450.4 mg, 2.50 mmol) and 1-chloro-2-iodoethane (952.1 mg, 5.00 mmol) in 1,4-dioxane (5.00 mL), was added potassium carbonate (380.1 mg, 2.75 mmol). The resulting suspension was then stirred and heated to 50 °C using microwave energy for 1 h. The reaction was allowed to cool to room temperature and then quenched with water, treated with 10% sodium thiosulfate to remove excess iodine and extracted into CHCl<sub>3</sub>. The organic solvent was then dried over Mg<sub>2</sub>SO<sub>4</sub>, filtered, and excess solvent was removed via rotary evaporation. The resulting material was purified by HPLC to generate the product **7** (410 mg, 1.7 mmol) in 67% yield. C<sub>9</sub>H<sub>11</sub>ClN<sub>4</sub>O<sub>2</sub>; HRMS (ESI) *m/z* calculated = 243.0643; found = 243.0654. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ 8.13 (s, 1H), 4.57 (t, *J* = 5.7 Hz, 2H), 4.03 (t, *J* = 5.7 Hz, 2H), 3.43 (s, 3H), 3.22 (s, 3H). <sup>13</sup>C NMR (125 MHz, DMSO-*d*<sub>6</sub>) δ 154.44, 150.93, 148.59, 143.25, 105.77, 47.74, 43.56, 29.46, 27.58.



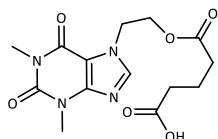
**1,3-dimethyl-7-(2-((1-phenylpropan-2-yl)amino)ethyl)-3,7-dihydro-1H-purine-2,6-dione (1).** To a microwave vessel containing a solution of **7** (350.0 mg, 1.00 mmol), and **2** (135.2 mg, 1.00 mmol) in DMF (5.00 mL) were added potassium iodide (332.0 mg, 2.00 mmol) and potassium carbonate (276.42 mg, 2.00 mmol). The resulting suspension was then stirred and heated to 150 °C using microwave energy for 20 min. The reaction was allowed to cool to room temperature, then quenched with water and extracted into CHCl<sub>3</sub>. The organic solvent was then dried over Mg<sub>2</sub>SO<sub>4</sub>, filtered, and excess solvent was removed via rotary evaporation. The resulting material was purified by HPLC to generate the product **1** (292 mg, 0.86 mmol) in 85% yield. C<sub>18</sub>H<sub>23</sub>N<sub>5</sub>O<sub>2</sub>; HRMS (ESI) *m/z* calculated = 342.1924; found = 342.1924. <sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 7.95 (s, 1H), 7.35 – 7.20 (m, 4H), 7.19 – 7.06 (m, 2H), 4.71 (t, *J* = 5.4 Hz, 2H), 3.82 – 3.58 (m, 3H), 3.58 (s, 3H), 3.38 (s, 3H), 3.11 (dd, *J* = 13.7, 5.8 Hz, 1H), 2.80 (dd, *J* = 13.7, 8.7 Hz, 1H), 1.32 (d, *J* = 6.5 Hz, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 156.03, 151.08, 148.74, 142.78, 134.68, 129.26, 129.19, 127.88, 106.77, 77.16, 57.30, 45.43, 44.27, 39.74, 30.46, 28.60, 16.05.



5-((2-(1,3-dimethyl-2,6-dioxo-1,2,3,6-tetrahydro-7H-purin-7-yl)ethyl)(1-phenylpropan-2-yl)amino)-5-oxopentanoic acid (FEN, **8**). To a round bottom flask containing a stirring solution of glutaric anhydride (45.6 mg, 0.40 mmol) in chloroform (4.00 mL), **1** (136.6 mg, 0.40 mmol) was added dropwise. The resulting solution was then stirred for 2 h at 22 °C. The excess solvent was then removed via rotary evaporation. The resulting material was purified by HPLC to generate the product **8** (77 mg, 0.17 mmol) in 42% yield. C<sub>23</sub>H<sub>29</sub>N<sub>5</sub>O<sub>5</sub>;

HRMS (ESI)  $m/z$  calculated = 456.2241; found = 456.2240. <sup>1</sup>H NMR (400 MHz, Methanol-*d*<sub>4</sub>) δ 7.26 – 7.14 (m, 5H), 4.51 – 4.38 (m, 2H), 4.24 (p,  $J$  = 7.2 Hz, 2H), 3.68 – 3.61 (m, 1H), 3.52 (d,  $J$  = 3.3 Hz, 3H), 3.38 (s, 3H), 2.79 (d,  $J$  = 7.3 Hz, 2H), 2.18 – 2.05 (m, 2H), 1.83 – 1.50 (m, 4H), 1.22 (d,  $J$  = 6.7 Hz, 3H). <sup>13</sup>C NMR (125 MHz, DMSO-*d*<sub>6</sub>) δ 175.10, 173.10, 155.38, 149.31, 143.54, 139.43, 129.96, 129.13, 127.18, 107.22, 54.72, 45.66, 33.77, 32.38, 30.32, 28.53, 20.91, 19.73.

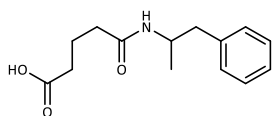
### Synthesis of THEO Hapten.



5-(2-(1,3-dimethyl-2,6-dioxo-1,2,3,6-tetrahydro-7H-purin-7-yl)ethoxy)-5-oxopentanoic acid (THEO, **9**). To a microwave vessel containing a solution of **6** (180.2 mg, 1.00 mmol) in DMF (2.00 mL) was added 2-bromoethan-1-ol (500 mg, 4.00 mmol) and sodium hydride (48.0 mg, 2.00 mmol). The resulting solution was then stirred and heated to 150 °C using microwave energy for 1 h. The reaction

was allowed to cool to room temperature, quenched with minimal H<sub>2</sub>O (40 μL), and the excess solvent was removed via rotary evaporation to generate the crude intermediate 7-(2-hydroxyethyl)-1,3-dimethyl-3,7-dihydro-1H-purine-2,6-dione (193 mg, 0.86 mmol), which was taken directly forward without any additional purification. A portion of this residue (56.1 mg, 0.25 mmol) was suspended in THF (2.00 mL) to which 4-dimethylaminopyridine (15.3 mg, 0.13 mmol) and glutaric anhydride (28.5 mg, 0.25 mmol) were added. The resulting solution was stirred and heated to 90 °C for 16 h. The excess solvent was then removed via rotary evaporation. The resulting material was purified by HPLC to generate the product **9** (28 mg, 0.08 mmol) in 33% yield. C<sub>14</sub>H<sub>18</sub>N<sub>4</sub>O<sub>6</sub>; HRMS (ESI)  $m/z$  calculated = 339.1299; found = 339.1297. <sup>1</sup>H NMR (500 MHz, DMSO-*d*<sub>6</sub>) δ 8.08 (s, 1H), 4.50 (d,  $J$  = 5.1 Hz, 2H), 4.39 (d,  $J$  = 5.1 Hz, 2H), 3.42 (s, 3H), 3.23 (s, 3H), 2.27 (t,  $J$  = 7.4 Hz, 2H), 2.17 (t,  $J$  = 7.4 Hz, 2H), 1.69 – 1.62 (m, 2H). <sup>13</sup>C NMR (125 MHz, DMSO-*d*<sub>6</sub>) δ 173.91, 172.10, 154.47, 150.98, 148.41, 142.95, 106.02, 62.30, 45.38, 32.53, 29.44, 27.56, 19.69.

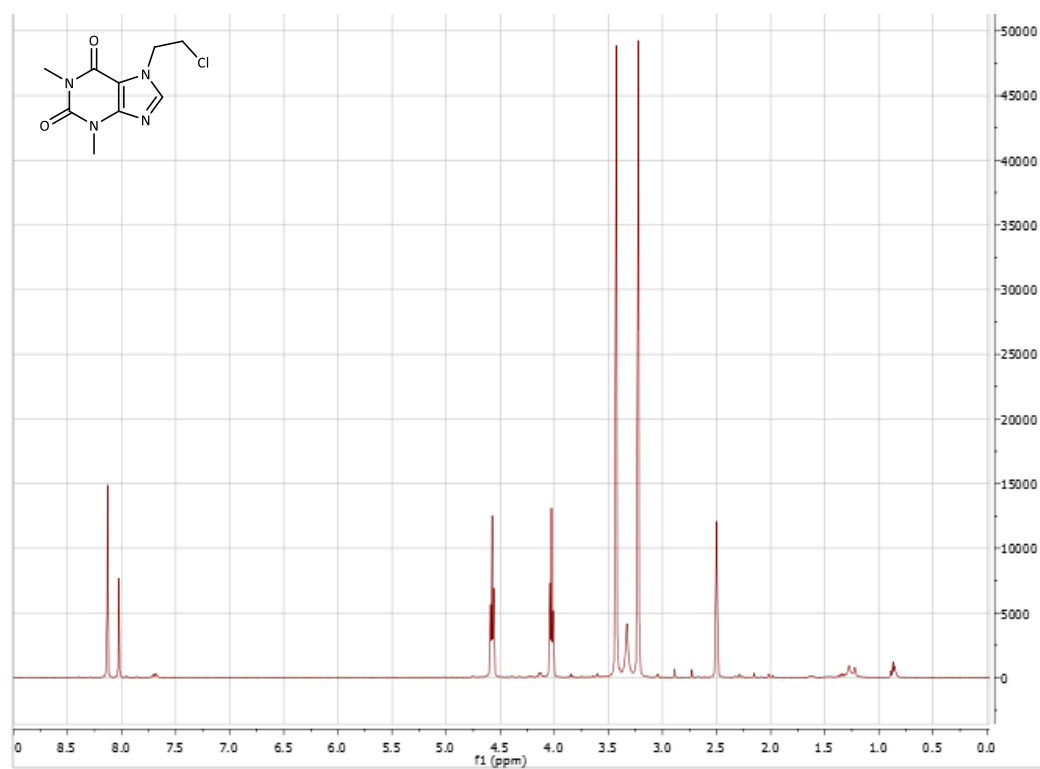
### Synthesis of AMPH Hapten.



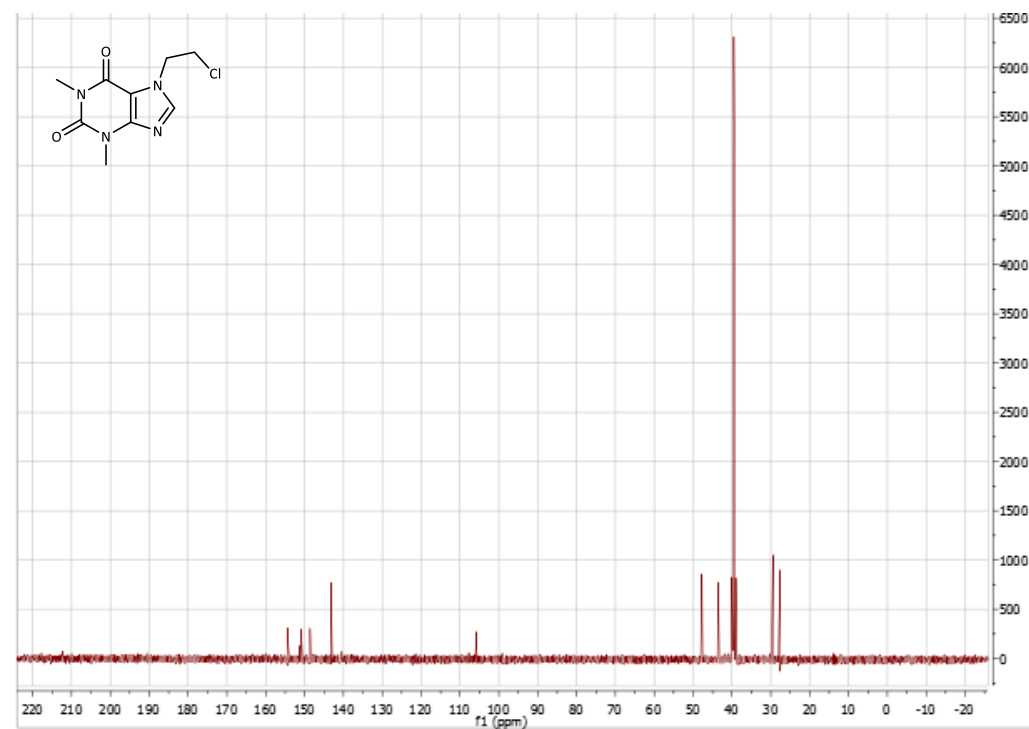
5-oxo-5-((1-phenylpropan-2-yl)amino)pentanoic acid (AMPH, **10**). To a microwave vessel containing a stirring solution of glutaric anhydride (45.6 mg, 0.40 mmol) in THF (4.00 mL), **2** (54.1 mg, 0.40 mmol) was added dropwise. The resulting solution was then stirred and heated to 90 °C using microwave energy for 1 h. The excess solvent was then removed via rotary evaporation.

The resulting material was purified by HPLC to generate the product **10** (44 mg, 0.18 mmol) in 44% yield. C<sub>14</sub>H<sub>19</sub>NO<sub>3</sub>; HRMS (ESI)  $m/z$  calculated = 250.1438; found = 250.1434. <sup>1</sup>H NMR (500 MHz, DMSO-*d*<sub>6</sub>) δ 7.72 (d,  $J$  = 8.1 Hz, 1H), 7.30 – 7.23 (m, 2H), 7.21 – 7.14 (m, 3H), 3.95 (dt,  $J$  = 14.2, 6.9 Hz, 1H), 2.71 (dd,  $J$  = 13.2, 7.1 Hz, 1H), 2.59 (dd,  $J$  = 13.3, 6.8 Hz, 1H), 2.12 (t,  $J$  = 7.4 Hz, 2H), 2.08 – 1.97 (m, 2H), 1.65 (p,  $J$  = 7.3 Hz, 2H), 1.00 (d,  $J$  = 6.6 Hz, 3H). <sup>13</sup>C NMR (125 MHz, DMSO-*d*<sub>6</sub>) δ 174.14, 170.63, 139.09, 129.07, 128.04, 125.93, 45.69, 41.87, 34.55, 32.93, 20.68, 20.14.

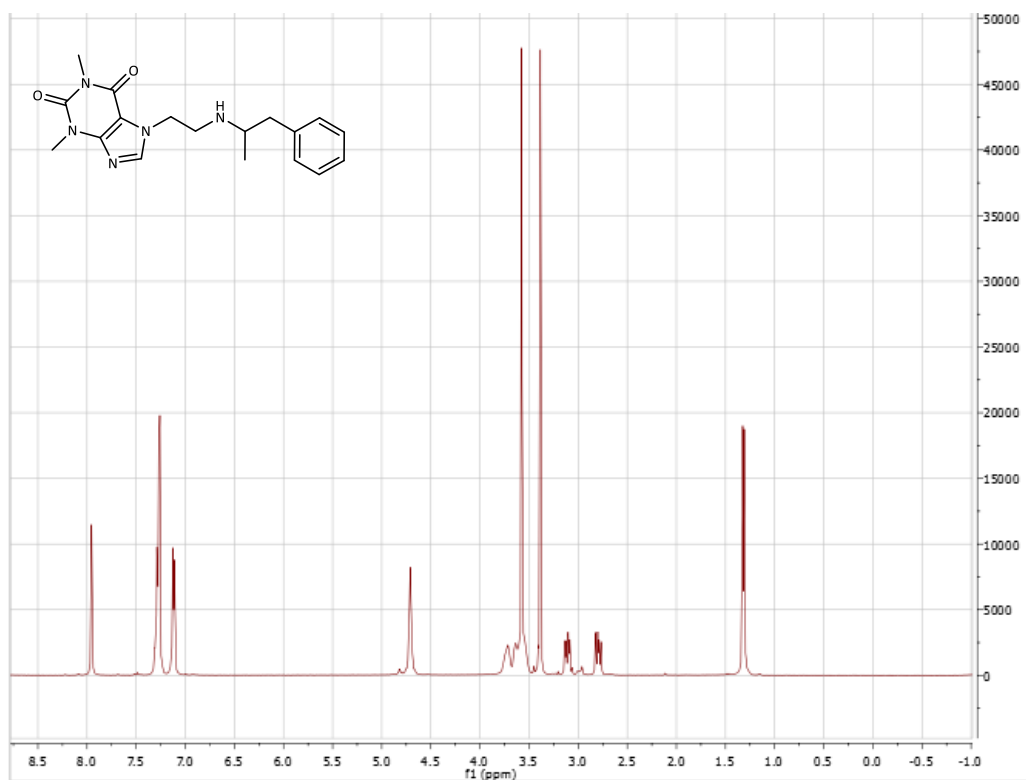
## NMR SPECTRA



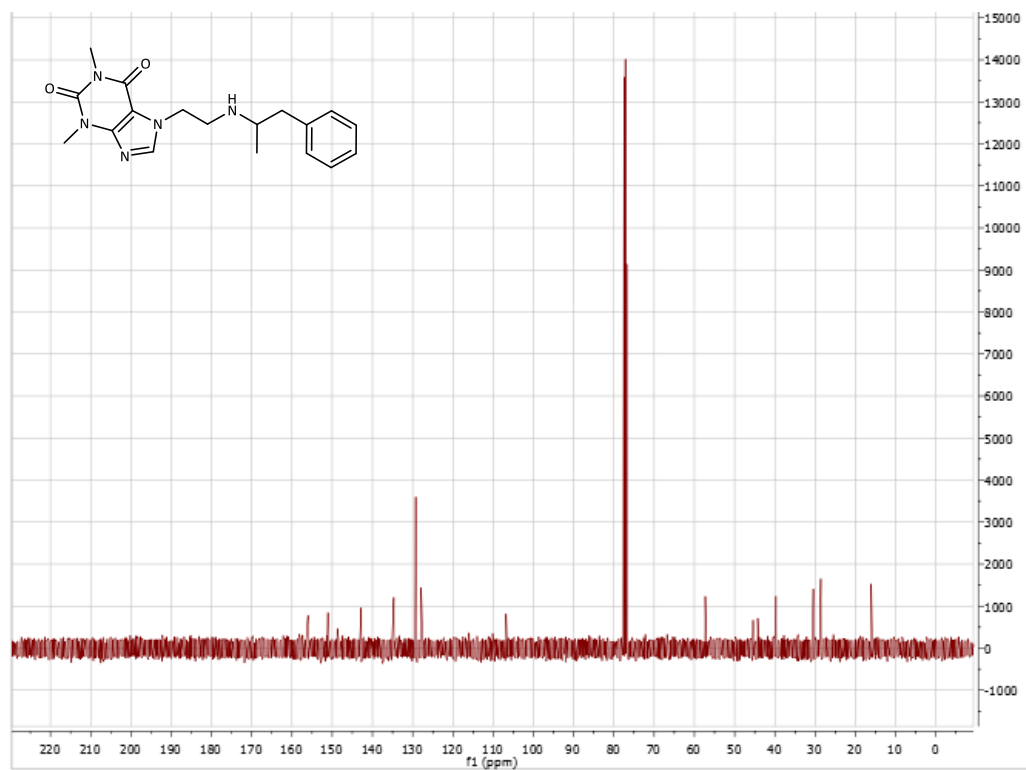
Supplementary Figure 1.  $^1\text{H}$  NMR for compound 7



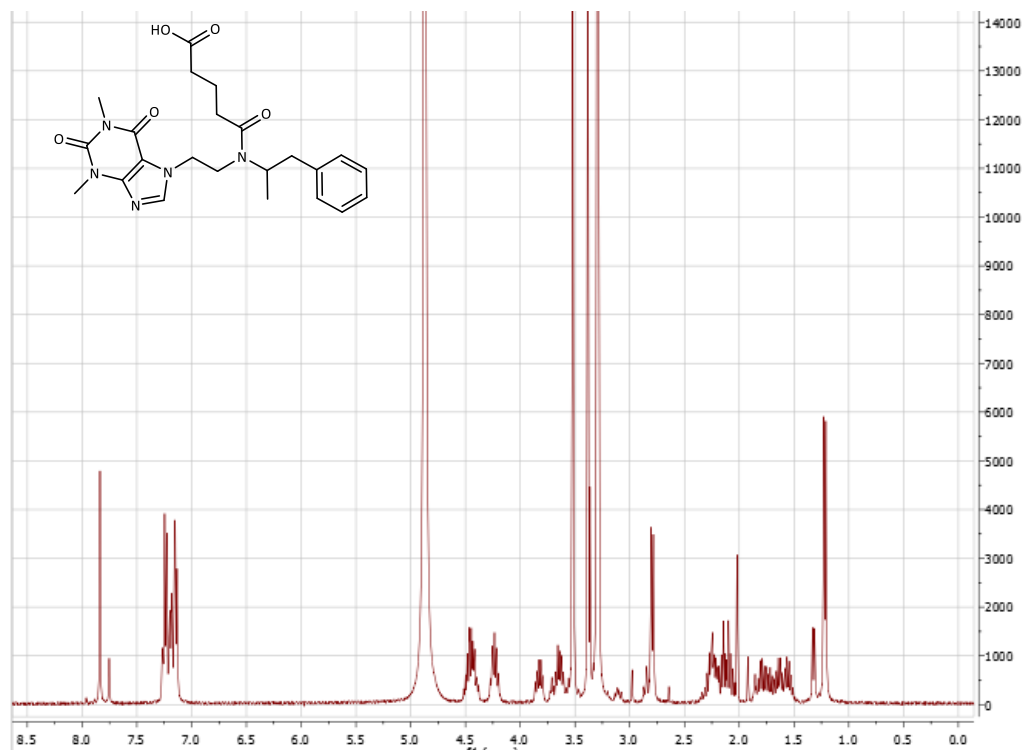
Supplementary Figure 2.  $^{13}\text{C}$  NMR for compound 7



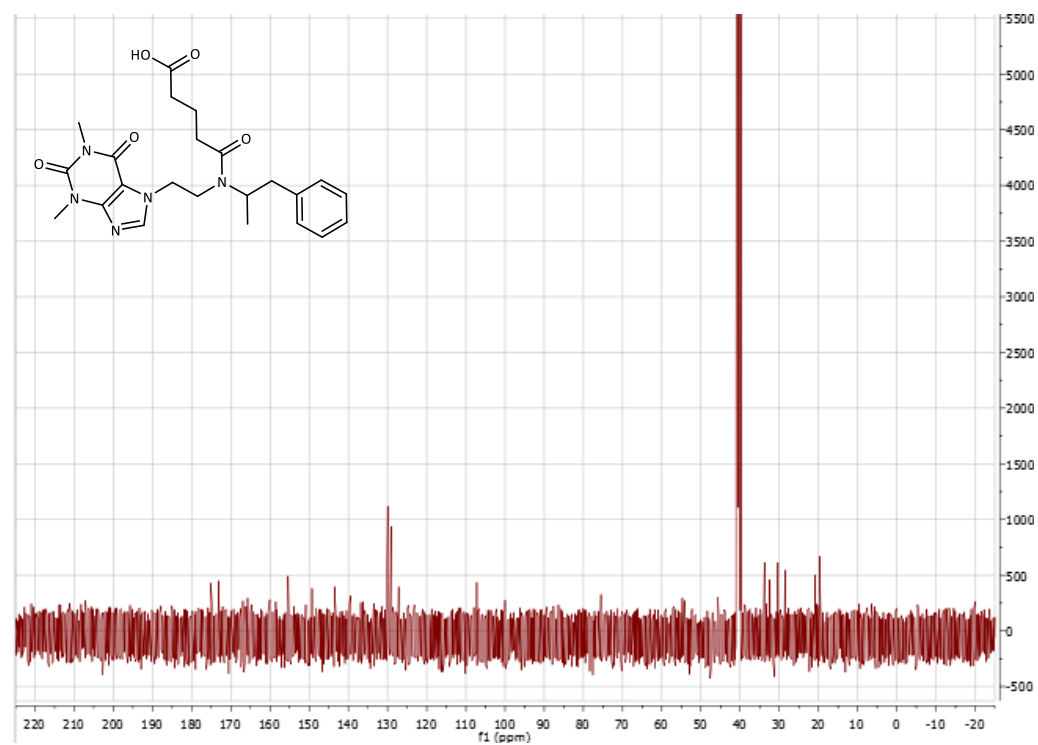
**Supplementary Figure 3.** <sup>1</sup>H NMR of compound 1



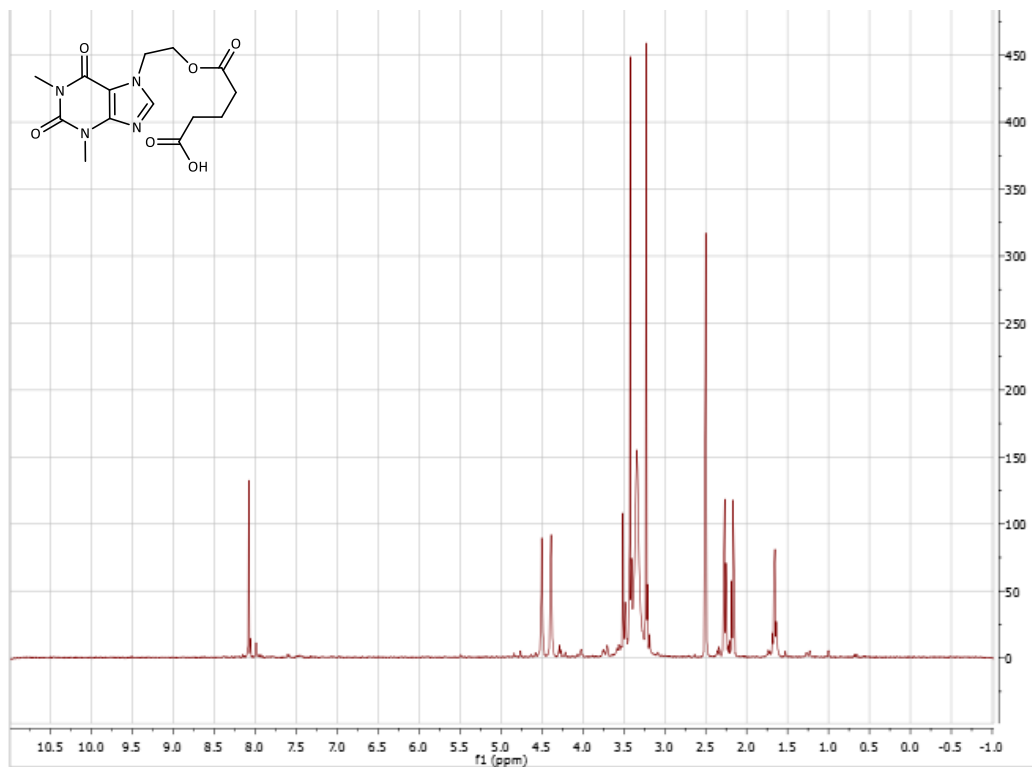
**Supplementary Figure 4.** <sup>13</sup>C NMR of compound 1



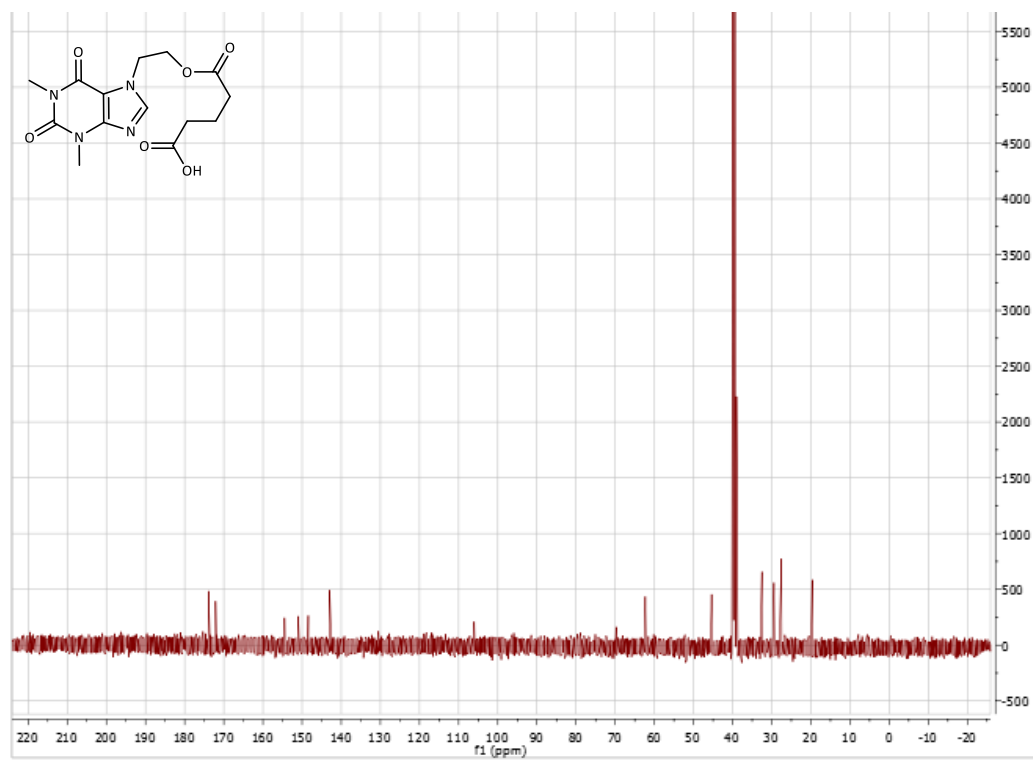
Supplementary Figure 5. <sup>1</sup>H NMR of compound 8



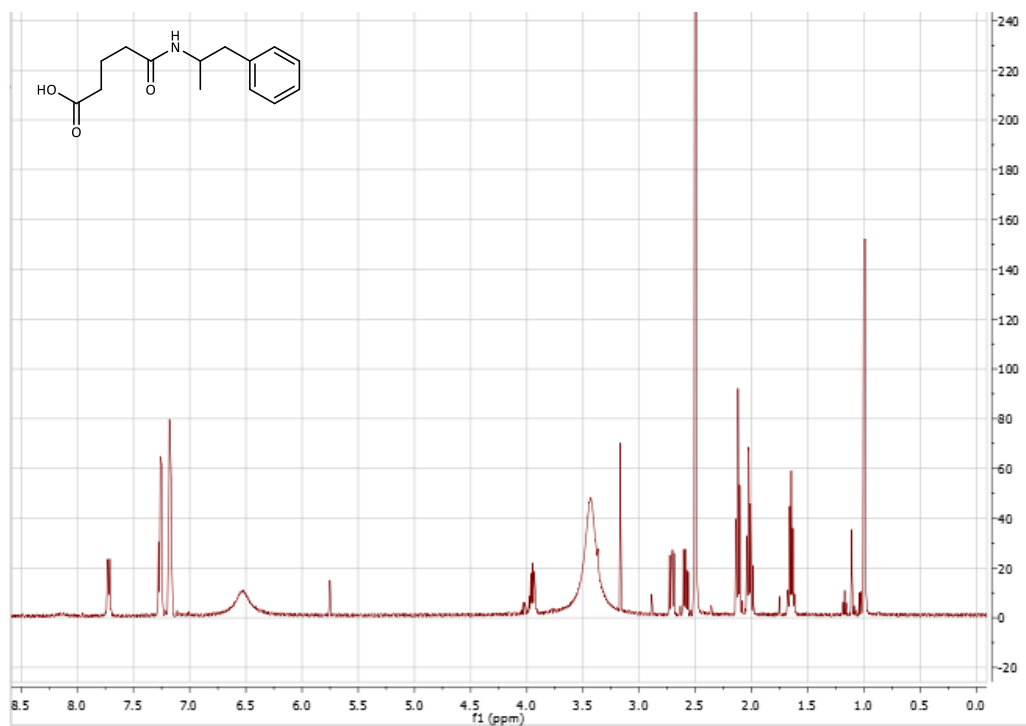
Supplementary Figure 6. <sup>13</sup>C NMR of compound 8



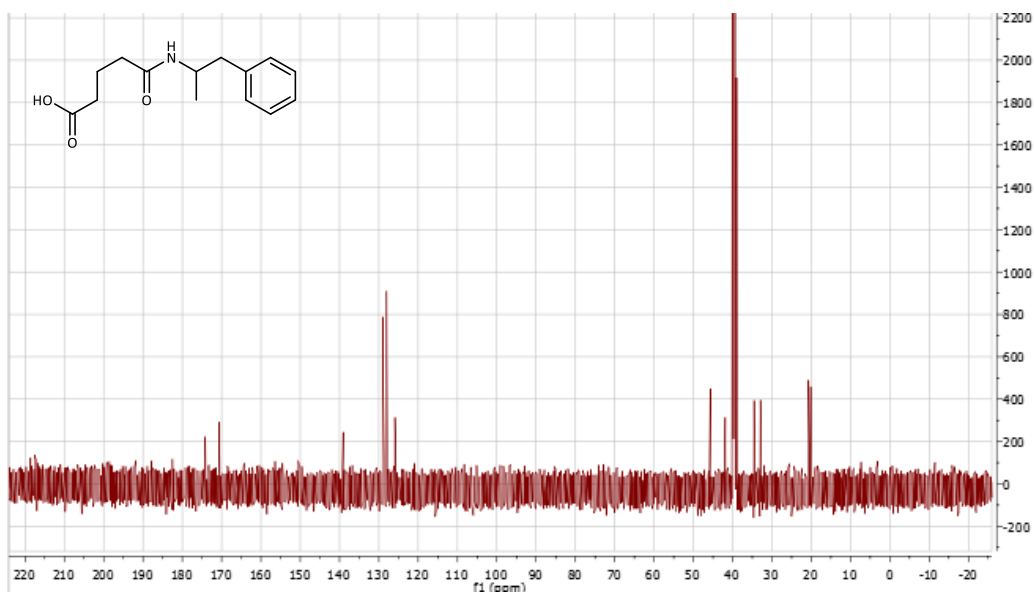
**Supplementary Figure 7.** <sup>1</sup>H NMR of compound 9



**Supplementary Figure 8.** <sup>13</sup>C NMR of compound 9



**Supplementary Figure 9.** <sup>1</sup>H NMR of compound 10



**Supplementary Figure 10.** <sup>13</sup>C NMR of compound 10